



Education and Sport Development

Department of Education and Sport Development

Departement van Onderwys en Sportontwikkeling

Lefapha la Thuto le Tlhabololo ya Metshameko

NORTH WEST PROVINCE

PROVINCIAL ASSESSMENT

GRADE 10

TECHNICAL SCIENCES P1

JUNE 2019

MARKS: 100

TIME: 2 hours

This question paper consists of 9 pages including 2 data sheets.



NW/JUNE/TEC-SCNE/ EMIS/6*****

INSTRUCTIONS AND INFORMATION

1. This question paper consists of FIVE questions. Answer ALL the questions in the ANSWER SCRIPT.
2. Start EACH question on a NEW page.
3. Number the answers correctly according to the numbering system used in this question paper.
4. Leave ONE line between two sub-questions, for example between QUESTION 2.1 and QUESTION 2.2.
5. You may use a non-programmable calculator.
6. You may use appropriate mathematical instruments.
7. Show ALL formulae and substitutions in ALL calculations.
8. Round off your final numerical answers to a minimum of TWO decimal places.
9. Give brief motivations, discussions, etcetera where required.
10. Write neatly and legibly.



QUESTION 1

Four options are provided as possible answers to the following questions. Each question has only ONE correct answer. Choose the answer and write only a letter next to the question number, for example 1.5C.

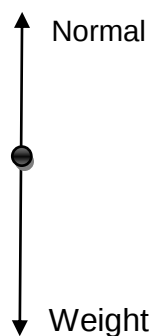
1.1 Which one of the following physical quantities is a scalar?

- A Time
 - B Velocity
 - C Displacement
 - D Acceleration
- (2)

1.2 Tension force exist when there is:

- A String
 - B Cable
 - C Rope
 - D All of the above
- (2)

1.3 A diagram below shows all the forces that are acting on a certain object.



According to the diagram above, the force applied on an object is:

- A To the right
 - B To the left
 - C Zero
 - D Equal to friction
- (2)

1.4 The product/s when hydrogen reacted with water is

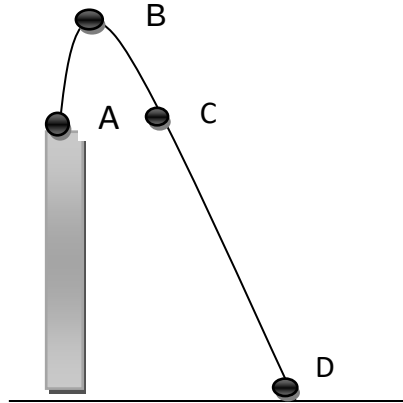
- A H_2O
 - B $\text{H} + \text{O}$
 - C HO_2
 - D $\text{H}_2 + \text{O}_2$
- (2)

1.5 The conversion of 1000 m in a scientific notation is

- A $1 \times 10^3 \text{ m}$
 - B $10 \times 10^2 \text{ m}$
 - C $0.1 \times 10^{-3} \text{ m}$
 - D $1 \times 10^{-3} \text{ m}$
- (2)



- 1.6 A stone from D was thrown to the top of the tank as shown in a diagram below. At what position is the magnitude of gravitational potential energy at minimum?



- A Position B
B Position C
C Position D
D Position A (2)
- 1.7 Moment is also known as:
A Force
B Torque
C Distance
D Fulcrum (2)
- 1.8 Acceleration is:
A The rate of velocity
B The rate of distance
C The rate of change in velocity
D The rate of displacement (2)
- 1.9 The kinetic energy of the boy with a mass of 48 kg running at a speed of 2.5m.s^{-1} will be:
A 120 J
B 150 J
C 300 J
D 60 J (2)



- 1.10. Two forces of magnitude $A = 220\text{N}$ and $B = 200\text{N}$ were applied as shown in the diagram below. The resultant of these force is:



- A 420 N left
- B 20 N left
- C 20 N right
- D 420 N right

(2)
[20]

QUESTION 2

- 2.1 Convert the following physical quantities and show all workings.

- 2.1.1 83°C to Fahrenheit. (3)
- 2.1.2 500 g to kilograms (2)
- 2.1.3 65 min to seconds (2)
- 2.1.4 The standard notation of 0.00000165 m to scientific notation. (2)

- 2.2 If Shane covers a distance of 0.5 km at a speed of 2.52 m/s. Calculate the time he took to complete the 0.5 km distance. (4)

- 2.3 If a car is moving at a speed of 60 km/h then changes to 85 km/h in 0.02 hours. Calculate the acceleration of the car. (5)
[18]

QUESTION 3

- 3.1 Define *pure substance*. (2)

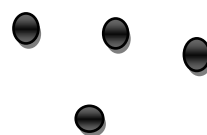
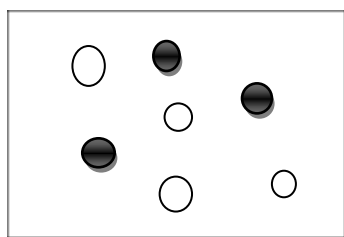
- 3.2 From the sketches below, choose:

A

B

C





- 3.2.1 Elements (1)
3.2.2 Compounds (2)

3.3 Consider the substances listed below.

silicon; potassium; zinc; sodium; sulphur; fluorine; bromine; boron

From the list above, write down the:

- 3.3.1 Metals (3)
3.3.2 Non-metals (3)
3.3.3 Metalloids (2)

3.4 Write down the NAMES of compounds formed from the combination of elements given below and their symbol.

3.4.1 Carbon and chlorine (2)

3.4.2 Magnesium and oxygen (2)

3.4.3 Lead and sulphur (2)

[19]

QUESTION 4

4.1 Differentiate between kinetic energy and gravitational potential energy. (4)

4.2 Define mechanical energy. (2)

4.3 An object of mass 8.5 kg is held 0.4 m above the table which is 1 m high. When the object was dropped it moved with a speed of 2 m.s^{-1} at height of 0.9 m above the floor.

4.3.1 Calculate the gravitational potential energy of the object relative to the table. (3)

4.3.2 What is the gravitational potential energy of the object relative to the floor? (4)

4.3.3 Calculate the kinetic energy of an object at 0.9 m above the floor. (3)

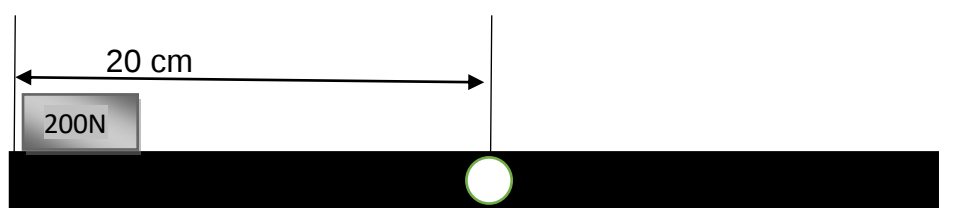
4.3.4 What is the mechanical energy of the object at 0.9 m above the floor? (4)

[20]

QUESTION 5

5.1 State the law of moment in words. (2)

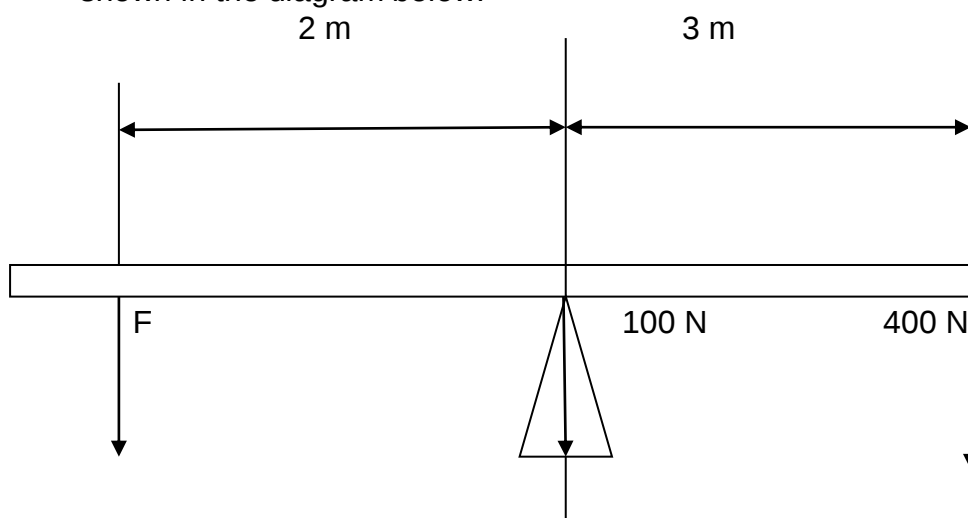
5.2 The weight of the block below is given as 200 N.



5.2.1 Determine the moment and write the direction. (4)

5.2.2 If another block with a weight of 300 N is put on the other side at the same distance. What is going to happen to a 200 N block? Take a guess and then prove it using calculation. (5)

5.3 The uniform beam in the diagram below is balanced when no loads are on it. The uniform beam weighs 100 N. Calculate the force F that must be applied to keep the beam in equilibrium, when a 400 N load is placed 3 m to the right as shown in the diagram below. (4)



5.4 Define mechanical advantage. (2)

5.5 If the length of the effort arm is 0.8 m and the load arm is 1.6 m. determine the mechanical advantage of the lever. (3)

5.6 If a load of 800 N is moved by a lever with a mechanical advantage of 1.5. What is the effort? (3)

[23]

GRAND TOTAL: 100

TABLE 1: PHYSICAL CONSTANTS

NAME	SYMBOL	VALUE
Acceleration due to gravity	g	9.8 m.s^{-2}
Charge on electron	e	$-1,6 \times 10^{-19}$



TABLE 2: FORMULEA

MOTION Speed = distance/ time Velocity = displacement/ time Acceleration = change in velocity/ time	ENERGY $E_p = mgh$ or $U = mgh$ $E_k = \frac{1}{2}mv^2$ $ME = E_p + E_k$
FORCE $F_g = mg$ $F_{res} = F_1 + F_2$	MOMENTS Moment = $F \times d$ OR Torque = $F \times d_{\perp}$
SIMPLE MACHINES $MA = \frac{L}{E}$ OR $MA = \frac{e}{l}$	



1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18
(I)	(II)											(III)	(IV)	(V)	(VI)	(VII)	(VIII)
1 2,1 H	3 7 Li	4 9 Be										5 11 B	6 12 C	7 14 N	8 16 O	9 19 F	10 20 Ne
	11 23 Na	12 24 Mg										13 27 Al	14 28 Si	15 31 P	16 32 S	17 35,5 Cl	18 40 Ar
	19 39 K	20 40 Ca	21 45 Sc	22 48 Ti	23 51 V	24 52 Cr	25 55 Mn	26 56 Fe	27 59 Co	28 59 Ni	29 63,5 Cu	30 65 Zn	31 70 Ga	32 73 Ge	33 75 As	34 79 Se	35 84 Br
	37 86 Rb	38 88 Sr	39 89 Y	40 91 Zr	41 92 Nb	42 96 Mo	43 96 Tc	44 101 Ru	45 103 Rh	46 106 Pd	47 108 Ag	48 112 Cd	49 115 In	50 119 Sn	51 122 Sb	52 128 Te	53 127 I
	55 133 Cs	56 137 Ba	57 139 La	72 179 Hf	73 181 Ta	74 184 W	75 186 Re	76 190 Os	77 192 Ir	78 195 Pt	79 197 Au	80 201 Hg	81 204 Tl	82 207 Pb	83 209 Bi	84 209 Po	85 210 At
	87 226 Fr	88 226 Ra	89 226 Ac														86 210 Rn

58 140 Ce	59 141 Pr	60 144 Nd	61 147 Pm	62 150 Sm	63 152 Eu	64 157 Gd	65 159 Tb	66 163 Dy	67 165 Ho	68 167 Er	69 169 Tm	70 173 Yb	71 175 Lu
90 232 Th	91 231 Pa	92 238 U	93 238 Np	94 238 Pu	95 238 Am	96 238 Cm	97 238 Bk	98 238 Cf	99 238 Es	100 238 Fm	101 238 Md	102 238 No	103 238 Lr

2 He	4 He
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Atomic number
Atoomgetal

Electronegativity
Elektronegatiwiteit

Symbol
Simbool

Approximate relative atomic mass
Benaderde relatiewe atoommassa

KEY/SLEUTEL

